



ALCHEMHY

ALTERNATIVE ROUTES FOR BASIC CHEMICALS PRODUCTION
USING HYDROGEN AS FEEDSTOCK

EU funded projects Clustering Workshop: Materials4Catalyst

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ALCHEMHY:

Alternative routes for basic chemicals
production using hydrogen as feedstock

Coordinator:

CIRCE, fundación circe centro de investigación de recursos y
consumos energeticos



ALCHEMHY

ALTERNATIVE ROUTES FOR BASIC CHEMICALS PRODUCTION
USING HYDROGEN AS FEEDSTOCK

**Reduce global emissions
18%**

**NET ZERO
2050**

Decarbonising
Europe's chemical backbone

green hydrogen as feedstock

**4 BREAK
TECHN**

**ALCHEMHY started
in October
2024!**

VIDEO!



SCAN CODE

INNOVATION

The Consortium - 16 partners; 8 EU countries

- **x7 Research institutions**

- CIRCE (Coordinator)
- Kemijski Institut
- **CIIAE** ←
- FBK – Fondazione Bruno Kessler
- AIT Austrian Institute of Technology
- CORE Innovation Centre
- ICCS

- **x3 Industries**

- UBE Corporation Europe
- Sonatrach Raffineria Italiana
- Casale

- **x2 Universities**

- Universidade de Aveiro
- Universiteit Antwerpen

- **x4 SMEs / Innovation organisations**

- Bluenergy Revolution
- White Research
- Hysytech
- Recatalyst



The Challenges

High emissions & energy use: The chemical industry is a major global CO₂ emitter, relying heavily on fossil-based raw materials.



Ammonia & methanol dependence: These essential chemicals are produced using hydrogen mainly derived from fossil fuels.



Big climate impact: Ammonia alone causes ~2% of global emissions, while methanol adds significantly through coal and natural gas use.



The Solution

Goal → Decarbonise ammonia & methanol production using renewable hydrogen.
→ Demonstrate 4 sustainable, cost-effective, fully electrified pathways

Methanol routes

- Small-flexible methanol reactor (SFMR)
- Plasma catalytic hydrogenation (PCH)

Ammonia routes

- Magnetic-heated sorption- enhanced reactor (MSER)
- Direct electrochemical synthesis (DESA)

Key features



Full **electrification**: induction, resistance, electrosynthesis, plasma.



Hybrid **digital twins** for design & operation (MSER, SFMR) and new catalyst materials for higher yields.



Industrial integration analysis: **technical, economic & environmental**.

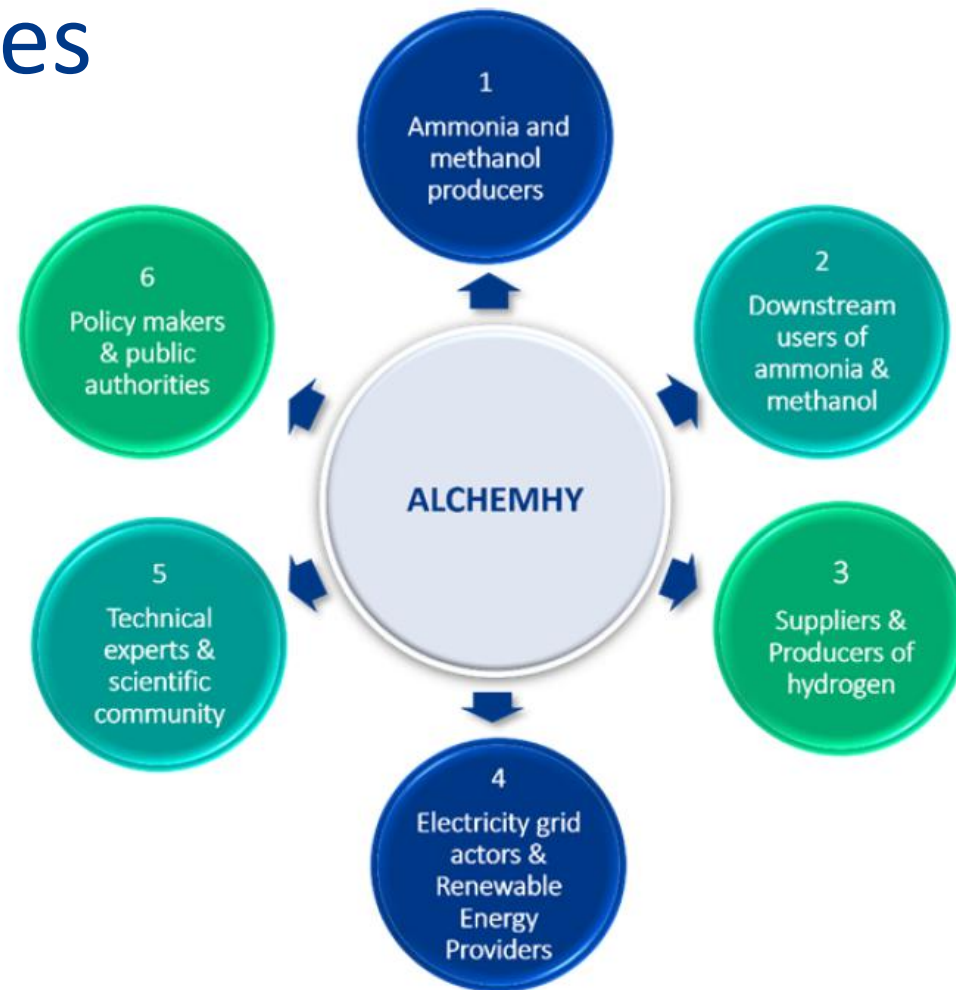


Replication, upscaling & business models for market adoption.

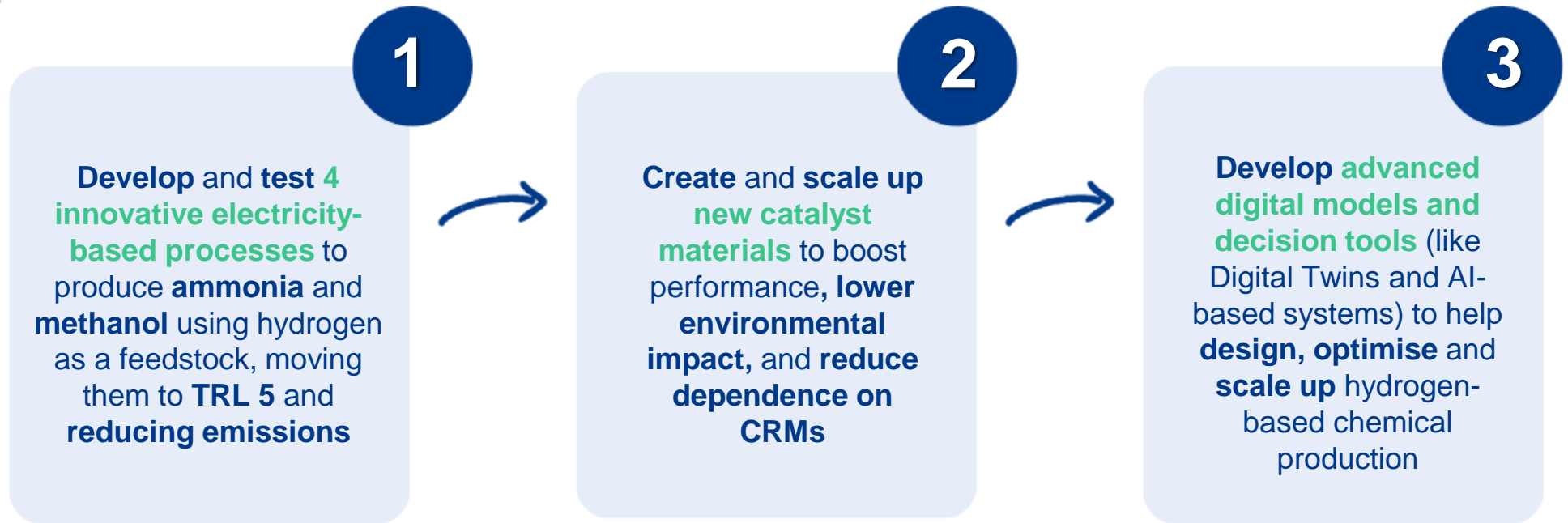


The Target audiences

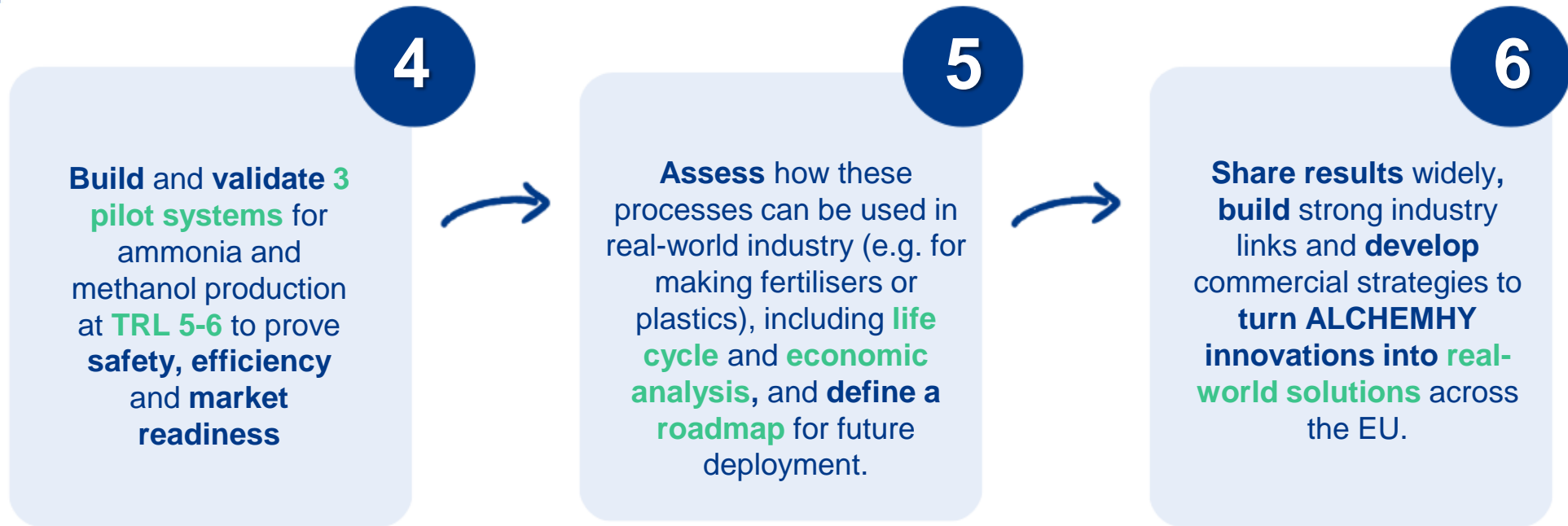
Who will benefit
from ALCHEMHY's
results?



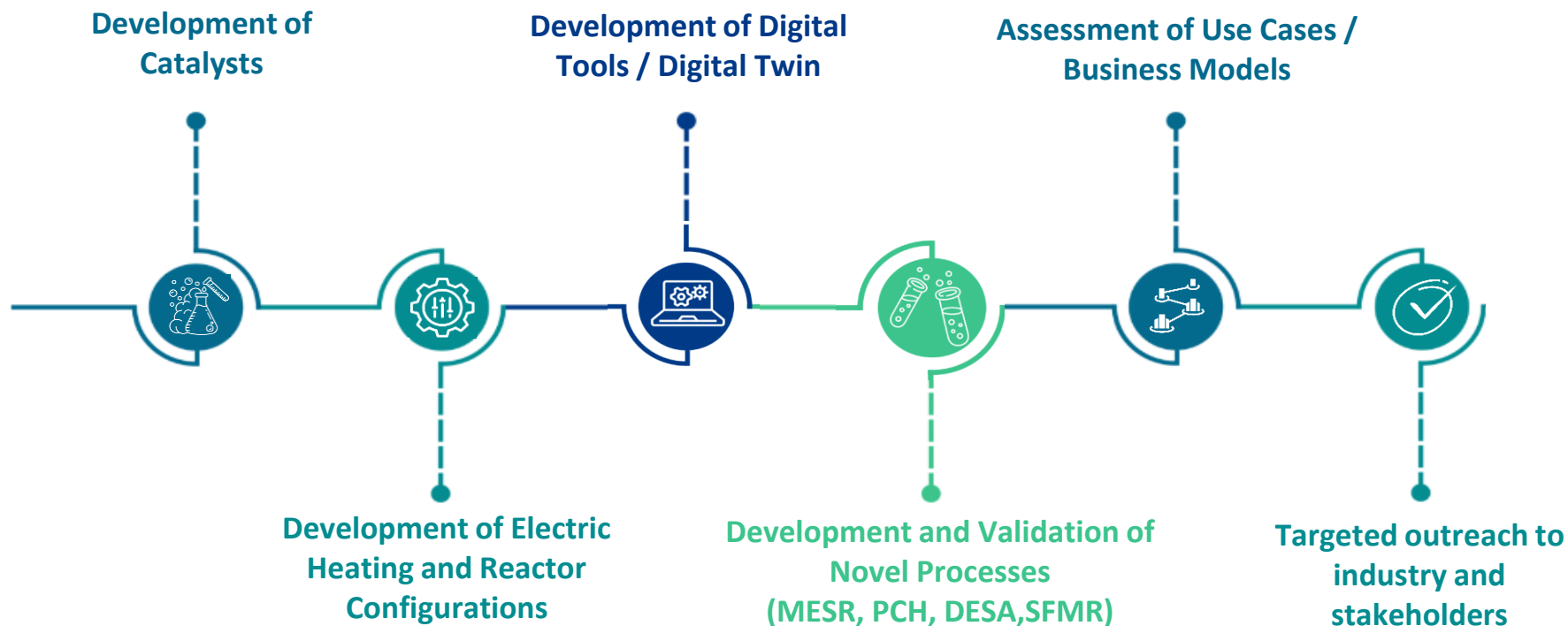
ALCHEMHY's objectives



ALCHEMHY's objectives



ALCHEMHY's key activities



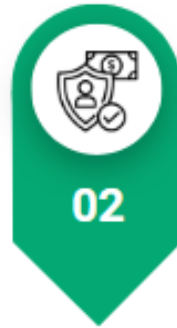
ALCHEMHY's impact



01

Technological & Scientific

- Proven H₂ feedstock solutions
- Scale-up & replication guidelines
- New process knowledge
- Synergy creation
- Smarter, more flexible process control



02

Market & Economic

- Less reliance on imported commodities
- Fossil fuel substitution
- Policy support with data
- Lower CAPEX needs
- New finance & business models



03

Societal & Environmental

- More renewables in industry
- Safer, cleaner processes
- Reduced CO₂ & pollutants
- Upskilled industrial workforce





**Reduce global emissions
18%**

**NET ZERO
2050**

Decarbonising
Europe's chemical backbone

green hydrogen as feedstock

**4 BREAKTHROUGH
TECHNOLOGIES**

**Ammonia
Methanol**

INNOVATION

**Real-world pilots
leading European labs**

Ammonia

MSER

TRL_{M0} → 4
TRL_{M48} → 6

Thermo-catalytic
process

Magnetic-heated
Sorption Enhanced
Reactor for flexible
ammonia synthesis

DESA

TRL_{M0} → 3
TRL_{M48} → 5-6

Thermo-Electro-
catalytic process

Direct Electrochemical
Synthesis
of Ammonia
in Solid Oxide Cells

Methanol

SFMR

TRL_{M0} → 4
TRL_{M48} → 6

Thermo-catalytic
process

Small, Flexible
Methanol Reactor
designed for fast
response
and modularity

PCH

TRL_{M0} → 3
TRL_{M48} → 5-6

Thermo-Plasma-
catalytic process

Plasma-Catalytic
Hydrogenation for
methanol production

Technologies for Methanol Production

- **Small-Flexible Methanol Reactor (SFMR)**

Description:

- Modular methanol synthesis on **high temperature** and under **relatively low-pressure** conditions
- **Fast response time** for dynamic RES
- Includes **Thermal Energy Storage (TES)** for optimal heat reuse

Key Features:

- **Flexible** to RES fluctuations
- **Modular, scalable, and retrofittable**
- **Containerised** for cost-effective deployment

Potential:

- Replicable in **CCU systems** (e.g. ethanol, DME)
- **Other:** catalyst/sorbent strategies adaptable to DME, olefins, ammonia





Technologies for Methanol Production

- **Small-Flexible Methanol Reactor (SFMR)**

Pilot testing:

- Construction of a **methanol synthesis** reactor able to **follow the RES fluctuation** and with a simplified design to improve its modularity and demonstrated at **BER facilities**.
- The focus is on demonstrating reliable **performance**, **safe operation**, and **scalability**



Technologies for Methanol Production

- **Plasma-Catalytic Hydrogenation (PCH)**

Description:

- Dielectric barrier discharge plasma reactor for **CO₂ hydrogenation**
- Operates at **room temperature and atmospheric pressure**
- Uses **catalysts** to boost methanol selectivity

Key Features:

- Thermodynamically favourable low-temperature conditions
- **Rapid on/off** switching for RES alignment
- Operates with **impure feeds and variable H₂**
- **Low CAPEX**

Potential:

- Suitable for **stable molecules** (N₂, CO₂, CH₄)
- **Different products**: syngas, fuels, olefins, ammonia
- Potential use in **other reactions** such as water-gas shift, hydrocarbon/ammonia cracking



University
of Antwerp



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Energético



CASALE



ReCatalyst

REvolutionising the way
we make fuel cell CATALYSTs.

Technologies for Methanol Production

- **Plasma-Catalytic Hydrogenation (PCH)**

Pilot testing:

- At **UANTWERPEN**, methanol production is demonstrated using plasma-catalytic and thermo-catalytic reactors
- The pilot validates **continuous operation, process optimization, and integration with green hydrogen** from PEM electrolyzers.



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Technologies for Ammonia Production

- **Magnetic-Heated Sorption-Enhanced Reactor (MSER)**

Description:

- Electrified ammonia synthesis via **induction heating and sorption**
- **Lower temperature and pressure** vs. Haber-Bosch
- **Validated at bench scale** in HySTrAm
- **Sorbents** offer high capacity, selectivity, and stability across variable temperatures

Key Features:

- **Fast ramp-up times**
- **Tuneable conversion:** from minimal to complete
- **Direct reactor heating** boosts energy efficiency
- **Reversible:** can crack ammonia into H_2 and N_2

Replication Potential:

- Use in **CO₂ hydrogenation**, bio-chemical **upgrading** (e.g. levulinic acid, furfural), and **ammonia cracking**



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we make fuel cell CATALYSTs.



CIIAE

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Energético

Technologies for Ammonia Production

- **Magnetic-Heated Sorption-Enhanced Reactor (MSER)**

Pilot testing:

- Pilot activities at **CIIAE** focus on validating lab-scale **ammonia production in a continuous process**
- The MESR is tested for efficiency, safety, and integration with green hydrogen, while data collected support digital **twin validation** and **process optimization**



Technologies for Ammonia Production

- **Direct Electrochemical Synthesis of Ammonia (DESA)**

Description:

- **Single-step ammonia synthesis** using hydrogen, nitrogen, and electricity
- High-temperature electrochemical route in **solid oxide cells (SOCs)**
- Operates at **400-450 °C with non-PGM catalysts**

Key Features:

- Achieves **high efficiency** using **waste heat**
- **Tolerant** to thermal/redox stress

Replication Potential:

- **Decentralised ammonia production** in agriculture, transport, and energy storage, **ammonia cracking**



Technologies for Ammonia Production

- **Direct Electrochemical Synthesis of Ammonia (DESA)**

Pilot testing:

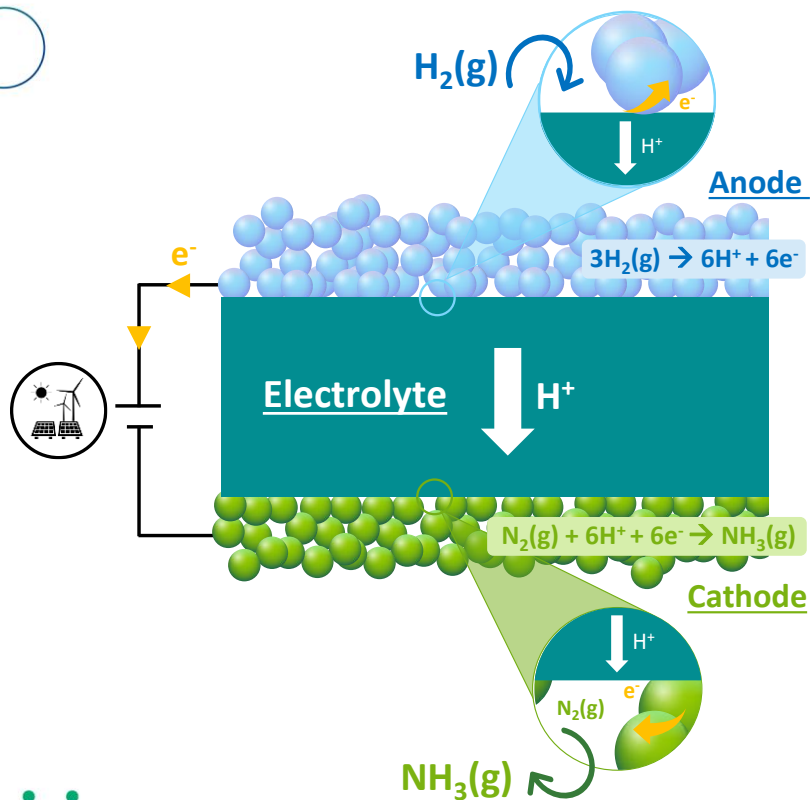
- Pilot testing at **CIIAE** will evaluate the DESA technology **under relevant operating conditions**.
- The focus is on demonstrating **process performance**, **safety**, and **operational stability**, while providing data for model validation and future scale-up.



Technologies for Ammonia Production

- Direct Electrochemical Synthesis of Ammonia (DESA)

DESA principles



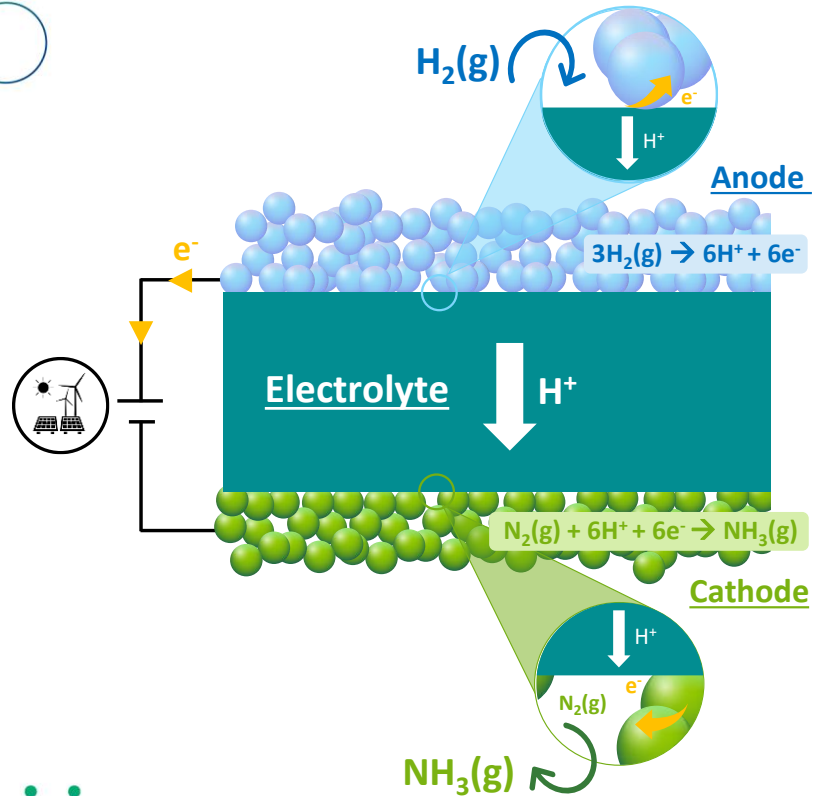
- Electrochemical cell based on proton-conducting ceramic materials
- Operation conditions:
 - 400-550 °C
 - 1 bar
- Key challenge: competing processes at the cathode side:
 - H_2 evolution reaction (HER)
 - Thermal decomposition of NH_3



Technologies for Ammonia Production

- Direct Electrochemical Synthesis of Ammonia (DESA)

DESA materials



- ✓ ANODE:

cermets \rightarrow Ni + electrolyte

- ✓ ELECTROLYTE:

perovskites + sintering aids (e.g. NiO, CuO, ZnO)

\rightarrow Ba(Ba,Zr,Y,Yb)O_{3-δ} (BCZYY)

- ✓ CATHODE:

electrolyte + perovskites + electrocat. (Ru, Fe, Co, Ni)

\rightarrow (Ba,Sr)(Co,Fe,Mo)O_{3-δ} (BSCFM)

\rightarrow Sr(Fe,Ti,Mo)O_{3-d} (SFT)

\rightarrow (Sr,La)TiO_{3-d} (SLT)

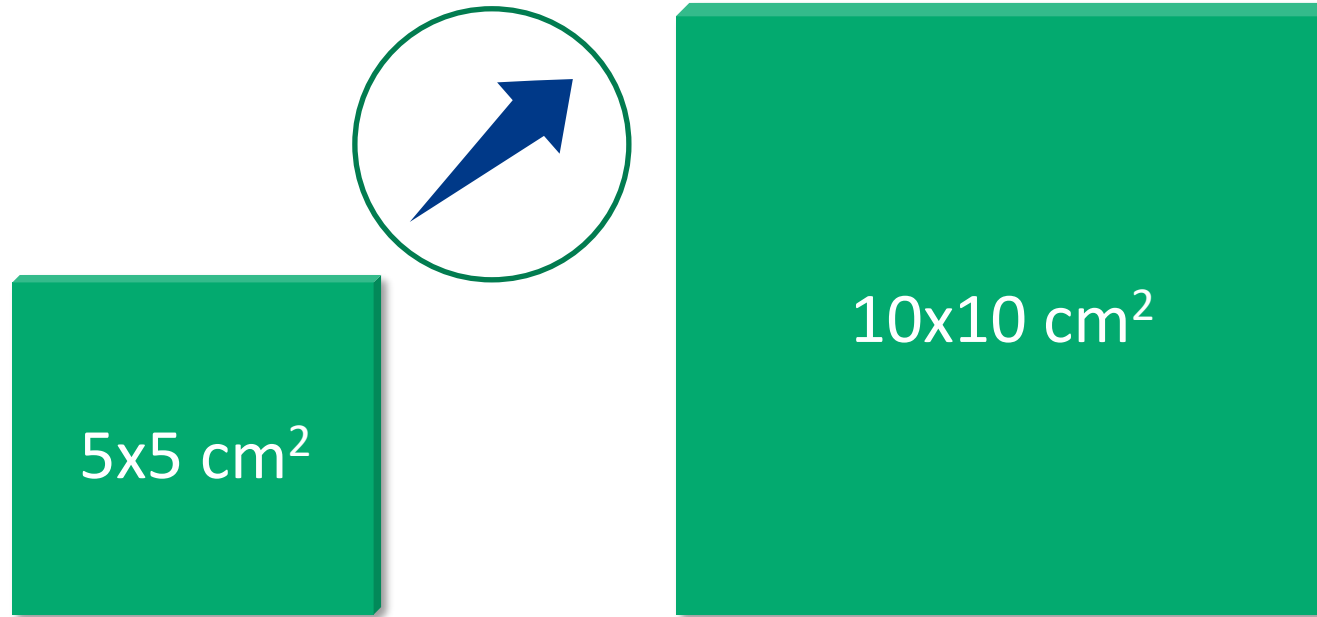
\rightarrow (La,Sr)_{1-x}(Cr,Mg)O_{3-d} (LSCM)

\rightarrow (La,Sr)(Fe,Co)O₄ (LSFC)

Technologies for Ammonia Production

- Direct Electrochemical Synthesis of Ammonia (DESA)

DESA scale





Thank you!



Visit: alchemhyproject.eu

Contact us: info@alchemhyproject.eu

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